

$$\frac{X \text{ mmol pdt}}{0.0375} = \frac{\frac{3.38}{4H}}{\frac{3}{3H}}$$

$$X = 0.0317 \text{ mmol} = 12.7 \text{ mg product}$$

$$\frac{(12.7 \text{ mg})}{13.04 \text{ mg}} * 100 = 97 \text{ wt\%}$$

